

Identification of c-di-GMP Derivatives Resistant to an EAL Domain Phosphodiesterase

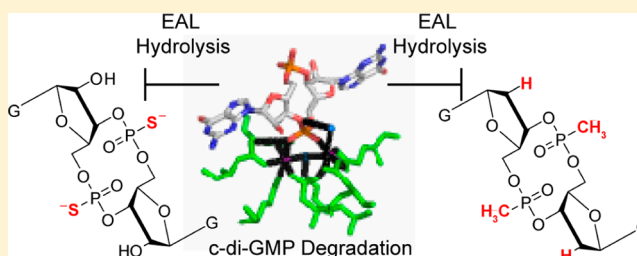
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S Supporting Information

ABSTRACT: The bacterial second messenger signaling molecule bis-(3'-5')-cyclic dimeric guanosine monophosphate (c-di-GMP) controls important biological processes such as biofilm formation, virulence response, and motility. This second messenger is sensed by macromolecular targets inside the cell, both protein and RNA, which induce specific phenotypic responses critical for bacterial survival. One class of enzymes responsible for regulating the intracellular concentration of c-di-GMP, and therefore the physiological behavior of the cell, consists of the EAL domain phosphodiesterases, which degrade the second messenger to its linear form, pGpG. Here, we investigate how base and backbone modifications of c-di-GMP affect the rate of cyclic dinucleotide degradation by an EAL domain protein (CC3396 from *Caulobacter crescentus*). The doubly substituted thiophosphate analogue is highly resistant to hydrolysis by this metabolizing enzyme but can still bind c-di-GMP riboswitch targets. We used these findings to develop a novel ribosyl phosphate-modified derivative of c-di-GMP containing 2'-deoxy and methylphosphonate substitutions that is charge neutral and demonstrate that this analogue is also resistant to EAL domain-catalyzed degradation. This suggests a general strategy for designing c-di-GMP derivatives with increased enzymatic stability that also possess desirable properties for development as chemical probes of c-di-GMP signaling.



Bacteria have the ability to sense different external signals present in diverse environments and translate these cues into physiological responses that can be critical for their survival.¹ One mechanism bacteria have evolved to facilitate such behavioral adaptation is the use of the second-messenger signaling molecule bis-(3'-5')-cyclic dimeric guanosine monophosphate (c-di-GMP) (for recent reviews, see refs 2–5). c-di-GMP is ubiquitous within the bacterial domain and is vital for regulating the transition between a sessile, biofilm-forming state and a motile, planktonic existence.⁶ This second messenger also controls the virulence response of pathogenic organisms⁷ and has been linked to quorum sensing, the process by which bacteria detect and communicate with one another.⁸

c-di-GMP is synthesized from two molecules of GTP by diguanylate cyclases (DGCs), which contain a GGDEF domain,⁹ and is degraded to the linear 5'-phosphate dinucleotide pGpG by EAL domain phosphodiesterases (PDEs) (Figure 1a).¹⁰ A second class of c-di-GMP phosphodiesterases known as the HD-GYP domain proteins have also been identified in some bacterial species, though these enzymes are much less common than the EAL domain proteins.^{11,12} GGDEF and EAL domain proteins are widely distributed throughout the bacterial domain,¹³ suggesting that this second messenger plays an essential biological role in many

different species. The opposing activities of these enzymes tightly regulate the concentration of c-di-GMP in the cell, which is directly sensed by downstream targets of the second messenger that act to induce the appropriate phenotypic response.^{2–5}

Although c-di-GMP is prevalent in bacteria, both the macromolecular targets that sense this second messenger and the specific mechanisms responsible for the observed behavioral transitions are only beginning to emerge and are highly diverse. Several classes of proteins, including PilZ domain proteins,¹⁴ transcription factors,^{15–17} and degenerate DGCs and PDEs,^{18–20} as well as two structurally distinct classes of riboswitch RNAs,^{21,22} have been identified as c-di-GMP effectors, underscoring the complexity of this key bacterial signaling pathway.

Understanding the metabolism of c-di-GMP is essential to elucidating how different bacterial phenotypes are triggered in response to varying intracellular concentrations of the second messenger. Structural characterization of this class of metabolizing enzymes indicates that c-di-GMP is bound in

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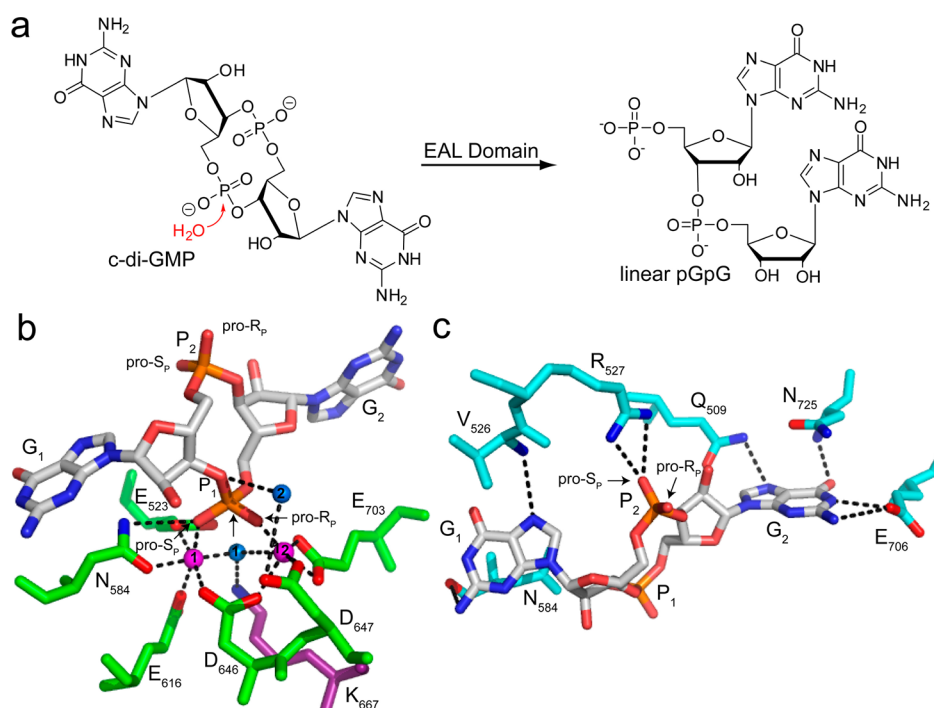


Figure 1. Recognition and enzymatic hydrolysis of c-di-GMP by EAL domain proteins. (a) Schematic reaction. In the presence of an EAL domain protein, a nucleophilic water molecule attacks one of the c-di-GMP phosphodiester bonds to yield the linear 5'-phosphate product, pGpG. (b) c-di-GMP bound to the active site of EAL domain protein TBD1265 from *Thiobacillus denitrificans* (PDB entry 3N3T). c-di-GMP is colored by atom: white for carbon, red for oxygen, blue for nitrogen, and orange for phosphorus. Water molecules are shown as blue spheres and magnesium ions as magenta spheres. The protein residues in the active site involved in metal coordination are colored green, and oxygens and nitrogens making contacts with c-di-GMP are colored red and blue, respectively. The highly conserved lysine that stabilizes the negative charge of water 1 and positions it for nucleophilic attack on the scissile phosphate, P₁, is colored purple. Water 2 is predicted to donate a proton to the leaving group. The *pro*-R_p and *pro*-S_p oxygens of P₁ and P₂ are labeled for sulfur substitution. (c) Recognition of the c-di-GMP bases and the phosphate distal to the site of catalysis, P₂ (PDB entry 3N3T). The coloring of c-di-GMP is the same as in panel b. Residues in contact with G1 and G2 are colored cyan, and the highly conserved arginine that makes electrostatic contacts with P₂ is colored white. Oxygens and nitrogens of protein side chains are colored as in panel b. Residues involved in G2 recognition are conserved, while those involved in G1 recognition are highly variable. The residue corresponding to N₇₂₅ is most often aromatic (F, W, or Y).

the extended conformation wherein the guanine bases (termed G1 and G2) are splayed apart from one another (Figure 1b,c),^{23–25} rather than the eclipsed conformation that is observed for the c-di-GMP binding riboswitches^{26,27} and many effector proteins.^{15,28,29} Few specific interactions are made to the guanine bases of the second messenger. The most highly conserved residues are involved in coordination of active site metals, the proposed nucleophilic water molecule, and the c-di-GMP phosphates (Figure 1b,c). Crystal structures of catalytically active EAL domain proteins reveal two metals in the active site that each contact a different oxygen atom of the scissile phosphate [P₁ (Figure 1b)], suggesting that hydrolysis of c-di-GMP is achieved via a two-metal ion mechanism and that both oxygens are critical to catalysis.^{23,25} Several inactive EAL domain proteins that can bind c-di-GMP but cannot catalyze degradation and instead function as downstream signaling molecules have been identified.^{19,30} These proteins contain mutations to the amino acids coordinating the metals, phosphate P₁, or the nucleophilic water molecule, highlighting the essential role these residues play in catalysis. The inactivity of such mutants suggests that changes to the c-di-GMP phosphate linkages may also impact reactivity.

Before the EAL domain was identified as the specific enzyme responsible for c-di-GMP degradation, Benziman and co-workers demonstrated that membrane extracts possessing c-di-GMP specific PDE activity were able to degrade doubly

substituted 2'-deoxy and inosine analogues and a singly substituted thiophosphate-modified analogue.³¹ This suggests that these particular modifications to c-di-GMP are unlikely to yield EAL domain resistant analogues, but these single-time point studies did not provide information about how these modifications affect the rate of second-messenger degradation or, in some cases, what effect double substitution has on activity. Therefore, we sought to determine which c-di-GMP modifications have the largest effect on catalysis by an EAL domain phosphodiesterase and identify those that render the second messenger resistant to degradation.

Here, we test a series of base-, ribose-, and phosphate-modified c-di-GMP derivatives to investigate the specificity of the EAL domain proteins for its cognate ligand. We identified modifications that render the second messenger highly resistant to EAL domain-catalyzed degradation and used these findings to develop a charge neutral, EAL domain resistant methylphosphonate c-di-GMP derivative.

■ MATERIALS AND METHODS

Materials. Radiolabeled c-di-GMP (*c-di-GMP) and unlabeled c-di-GMP were synthesized enzymatically using purified PleD* and tDGC diguanylate cyclase proteins, respectively, as previously described.^{32,33} Base- and ribose-modified analogues were synthesized on solid phase as previously described.³⁴ Mono- and dithiophosphate analogues

were synthesized in solution as previously described.³⁵ The absolute stereochemistry of the thiophosphate c-di-GMP analogues was assigned as previously described on the basis of the specificity of P1 and venom phosphodiesterases.³⁶ 3'-Phosphate CPG beads, guanosine methylphosphonamidite precursors, and Poly-Pak II desalting columns were purchased from Glen Research. 4-(Dimethylamino)pyridine (DMAP), tetrahydrofuran (THF), acetic anhydride, anhydrous pyridine, anhydrous acetonitrile (ACN), triethylamine (TEA), and 1-(2-mesitylenesulfonyl)-3-nitro-1*H*-1,2,4-triazole (MSNT) were purchased from Sigma. The 100-nucleotide class I Vc2 riboswitch RNA from *Vibrio cholerae* containing 2-aminopurine at position 94 [G94(2AP) RNA] was prepared as previously described.³⁴ The wild-type class II aptamer from *Clostridium acetobutylicum* was cloned and transcribed in vitro using T7 RNA polymerase as previously described.^{27,34}

Expression and Purification of Phosphodiesterase Protein CC3396 from *Caulobacter crescentus*. Six-histidine-tagged EAL domain phosphodiesterase protein CC3396 was purified by affinity chromatography using Ni-NTA agarose (Qiagen) as previously described.³⁷ Overnight cultures of *Escherichia coli* BL21 cells harboring the expression plasmid (pET21) encoding CC3396 were grown at 37 °C. The overnight culture was diluted into fresh LB medium, and cells were grown to an OD₆₀₀ of 0.6. Protein expression was then induced for 45 min at 37 °C via addition of 0.5 mM isopropyl 1-thio-β-D-galactopyranoside (IPTG). Cells were pelleted, resuspended in lysis buffer [50 mM sodium phosphate (pH 7.0), 300 mM NaCl, and 5 mM BME], and lysed by being passed through a microfluidizer at 15000 psi. The lysate was cleared by centrifugation (20000g), and the cleared lysate was incubated with Ni-NTA agarose for 1 h at 4 °C to allow the protein to bind to the column. The column was washed with lysis buffer containing 20 mM imidazole and then washed with lysis buffer containing 50 mM imidazole. The protein was eluted from the column in 250 mM imidazole, and the pure protein fractions were pooled and dialyzed into 25 mM Tris-HCl (pH 8.0), 250 mM NaCl, and 5 mM BME. Dialyzed protein was concentrated and stored in 10% glycerol.

Phosphodiesterase Assays for Evaluating Degradation Rates of c-di-GMP and Its Analogues. Phosphodiesterase activity was evaluated under multiple-turnover conditions with saturating substrate concentrations. Pure CC3396 protein (1 μM) was incubated with 100 μM GTP and 100 μM c-di-GMP (or an analogue) in 1× PDE buffer [10 mM MgCl₂, 25 mM Tris-HCl (pH 8.0), and 250 mM NaCl] at room temperature (22 °C). Aliquots were removed at each time point, and the reaction was quenched when the mixture was heated to 95 °C for 5 min. Degradation was monitored in the range where the reaction was still linear to measure the initial velocity. Samples were diluted in 50 mM TEAAc buffer (pH 6.0), and compound degradation was analyzed by high-performance liquid chromatography (HPLC) on a reverse phase C8 column using a gradient from 0 to 4% ACN in 50 mM TEAAc (pH 6.0) over 15 min for most compounds. A steeper gradient was used for the following compounds as indicated: 0 to 15% ACN for c-(R_pR_p)-di-G_{ps}, 0 to 10% ACN for c-(R_pS_p)-di-G_{ps}, 0 to 10% ACN for c-(R_p)-G_{ps}-GMP, and 0 to 20% ACN for c-di-dG_{p(Me)}-I and c-di-dG_{p(Me)}-II. The area under peaks corresponding to linear and cyclic compounds was determined by integration (HPChem Software), and the fraction of linear (F_L) product formed was calculated using the following equation:

$$F_L = \frac{\text{area}_L}{\text{area}_L + \text{area}_C}$$

where area_L is the area of the linear product and area_C is the area of the cyclic compound. F_L was multiplied by the substrate concentration to obtain the amount of product formed, which was plotted against time. The data were fit to a line, and the slope of the line was used to determine the initial rate (k_{cat}).

Competition Phosphodiesterase Assays. The degradation of radiolabeled c-di-GMP (*c-di-GMP) to pGpG was monitored by determining the fraction of pGpG formed by separating cyclic and linear products by polyethyleneimine-cellulose thin layer chromatography (PEI-cellulose TLC). For the competition assays, CC3396 (10 nM) was incubated at room temperature for 5 min with GTP (100 μM) in 1× PDE buffer before the simultaneous addition of trace amounts of radiolabeled c-di-GMP and an excess of unlabeled competitor analogue (100 μM). Aliquots were removed at each time point and reactions quenched by the addition of an equal volume of 0.5 M EDTA. Reaction mixtures were extracted with a phenol/chloroform/isoamyl alcohol mixture (25:24:1) to remove the protein before analysis by PEI-cellulose TLC. PEI-cellulose TLC plates were run in a 1:1.5 (v/v) mixture of saturated NH₄SO₄ and 1.5 M KH₂PO₄ (pH 3.6), and the amount of c-di-GMP and pGpG was quantified as previously described.³⁷

Chemical Synthesis of the Methylphosphonate Analogue. The methylphosphonate c-di-GMP analogue was synthesized on a solid support (3'-phosphate CPG beads) on a 1 μmol scale as previously described³⁸ with the following adaptations (Figure 2a). The 5'-dimethoxytrityl (DMTr)-protected guanosine methylphosphonamidite (50 mM in THF) was coupled to the solid support using 5-benzyl mercaptotetrazole (125 mM in ACN) as the activator. Oxidation was performed using a 0.02 M solution of iodine in a THF/pyridine/water mixture (88:10:2), and unreacted sites were capped using a 1:1 mixture of 6.5% DMAP in THF and 10% acetic anhydride in THF. The 5'-DMTr group was removed using 3% dichloroacetic acid in dichloromethane followed by coupling, oxidation, capping, and detritylation of the second nucleotide as described above. The linear dinucleotide was cleaved from the solid support via incubation with a 10% TEA/ACN mixture and coevaporated with anhydrous pyridine. The dinucleotide was cyclized in solution using 0.1 M MSNT in pyridine under argon for approximately 96 h. Global deprotection was afforded by evaporating the cyclization solution and resuspending the dry product in a 45:45:10 ACN/EtOH/NH₄OH mixture. After incubation for 30 min at room temperature in the deprotection mixture, an equal volume of ethylenediamine was added and the room-temperature incubation was allowed to proceed for an additional 6 h. The mixture was diluted with water, and the pH was adjusted to 7.0 using 6 M HCl. The product was desalted on a Poly-Pak II column (Glen Research) according to the manufacturer's instructions. The desalted product was lyophilized to dryness and then purified by HPLC on a C18 reverse phase column using a gradient of 0 to 30% acetonitrile in 50 mM TEAAc (pH 6.0) over 30 min (Figure 2b). Two of the three possible diastereomers were synthesized, and the identity of the product was confirmed by ESI-MS in negative ion and positive ion mode (exact mass of 654.15; observed mass for c-di-dG_{p(Me)}-I [M - H]/1 = 653.144, [M + H]/1 = 655.1473, [M + Na]/1 = 677.1346; observed mass for c-di-dG_{p(Me)}-II [M - H]/1 = 653.1306, [M + H]/1 = 655.1335, [M

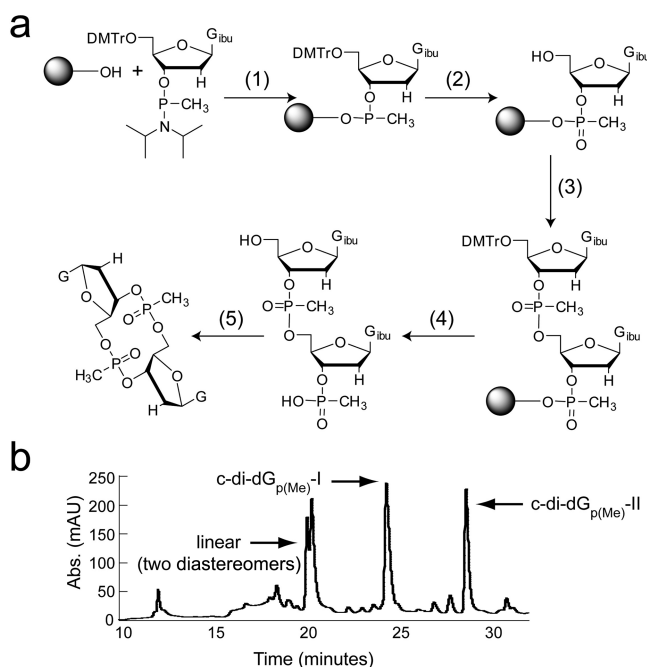


Figure 2. Synthesis of the double methylphosphonate c-di-GMP analogue. (a) Synthetic scheme. The linear dinucleotide was synthesized on the solid support and then cyclized in solution: (1) tetrazole/ACN; (2) (i) 3% DCA/DCM, (ii) I_2 /THF/pyridine, (iii) acetic anhydride/DMAPI, (iv) 3% DCA/DCM; (3) tetrazole/ACN with guanosine methylphosphonamidite/THF; (4) (i) 3% DCA/DCM, (ii) 10% TEA/ACN; (5) (i) 0.1 M MSNT/pyridine, (ii) ACN/ $EtOH/NH_4OH$, (iii) ethylenediamine. (b) HPLC trace monitored at 254 nm showing purification of the crude reaction synthesis for the methylphosphonate c-di-GMP analogue c-di-dG_{p(Me)}. Peaks corresponding to the two cyclized diastereomers and the linear, uncyclized molecule are labeled.

+ Na]/1 = 677.1207). The major impurity present is the noncyclized linear intermediate (Figure 2b). On the basis of the integration of peaks corresponding to the linear and cyclic compounds, the cyclization reaction proceeded with approximately 50% efficiency. The overall yield of the cyclic compound was 10% [7% c-di-dG_{p(Me)}-I and 3% c-di-dG_{p(Me)}-II].

Affinity Measurements of the Methylphosphonate Analogue for c-di-GMP Binding Riboswitches. Dissociation constants (K_d) of c-di-GMP analogues for the class I and class II riboswitches were determined as previously described by a competition gel shift assay with radiolabeled c-di-GMP.³⁴ Radiolabeled c-di-GMP was incubated with riboswitch aptamer RNA [25 nM class I [G94(2AP) variant] or 50 nM class II] and increasing concentrations of the analogue in folding buffer containing 10 mM MgCl₂, 10 mM NaCl, and 10 mM sodium cacodylate (pH 6.8) until equilibrium was achieved. Free c-di-GMP and RNA-bound c-di-GMP were separated by native PAGE [100 mM Tris/HEPES (pH 7.5) and 10 mM MgCl₂] at 4 °C. The amounts of free and bound c-di-GMP were quantified and fit to an equation for competitive binding as previously described to determine the K_d of the competitor analogue.³⁴

RESULTS

Monitoring the Enzymatic Hydrolysis of c-di-GMP and Its Analogues by HPLC. We employed an HPLC-based assay

to measure the rate of hydrolysis for c-di-GMP and its analogues by the EAL domain PDE CC3396 from *C. crescentus*. CC3396 is a prototypical PDE, and the activity of this protein has been well characterized in vitro.³⁷ It is a GGDEF-EAL protein containing a degenerate GGDEF domain that possesses no diguanylate cyclase activity but instead acts to allosterically activate the phosphodiesterase by binding to GTP.³⁷ Sequence alignment with other structurally and biochemically characterized PDEs indicates that the most highly conserved residues necessary for catalysis are conserved in this protein (Table S1 of the Supporting Information).^{23,25,39} Using this assay, we measured the rate of cyclic dinucleotide degradation for the singly substituted [c-GMP-IMP, c-(R_p)-G_{ps}-GMP, c-(S_p)-G_{ps}-GMP, and c-dG-GMP] and doubly substituted analogues (c-di-IMP, c-di-AMP, and c-di-dGMP) previously tested in single-time point studies.³¹ We also tested analogues whose susceptibility to EAL domain-catalyzed degradation had not been previously investigated [c-GMP-AMP, c-N¹mG-GMP, c-di-2'-F-GMP, c-di-2'-OMe-MP, c-(R_pR_p)-di-G_{ps}, c-(R_pS_p)-di-G_{ps}, and c-di-dG_{p(Me)}] (Figure 3). With the exception of c-di-IMP, c-di-AMP, and c-di-dGMP, the previous biochemical work looked only at the degradation of singly substituted analogues. Because c-di-GMP is a symmetric molecule, single substitutions are likely to leave the dinucleotide susceptible to degradation by targeting the unsubstituted position.

The rate of c-di-GMP degradation was measured under saturating conditions of both the second messenger [100 μ M (see below)] and GTP (100 μ M), and the conversion of c-di-GMP to pGpG was detected via HPLC by monitoring the appearance of a peak corresponding to pGpG (retention time of 7.5 min) and the disappearance of the peak corresponding to c-di-GMP (retention time of 10 min) (Figure 4a). Co-injection of the reaction mixture with a chemically synthesized standard of pGpG confirmed that the peak appearing at 7.5 min was the correct breakdown product. Under these conditions, the second messenger was completely degraded within 10 min, with the rate of c-di-GMP hydrolysis measured to be 15 mol min⁻¹ (mol of enzyme)⁻¹ (Figure 4 and Table 1). To confirm that we were using saturating concentrations of c-di-GMP, we measured the rate of degradation using 200 μ M c-di-GMP and found it to be the same as that measured using 100 μ M c-di-GMP [15.3 mol min⁻¹ (mol of enzyme)⁻¹ at 200 μ M c-di-GMP]. This indicates that the enzyme is saturated with substrate and that the observed rate corresponds to the k_{cat} . This was also confirmed for the ribosyl phosphate-modified analogues that were susceptible to degradation (Table 1), indicating that all rates measured for these analogues were determined under saturating conditions. The rate of hydrolysis of c-di-GMP by CC3396 that we measured is in reasonable agreement with that previously reported [115 mol min⁻¹ (mol of enzyme)⁻¹ at 30 °C],³⁷ and the different temperatures (37 °C vs 30 °C) and methods used to conduct the two studies likely account for the 8-fold difference in measurements.

Enzymatic Hydrolysis of Base-Modified Analogues.

Previous biochemical studies have demonstrated that mutations to EAL domain residues contacting the c-di-GMP bases do not significantly affect enzymatic activity.^{25,39} This suggests that modifications to the c-di-GMP bases would not be predicted to render the second messenger resistant to enzymatic degradation. We first tested the effects of modifying the c-di-GMP guanine bases with inosine, N¹-methylguanine, and adenine substitutions on the hydrolysis by EAL domain protein

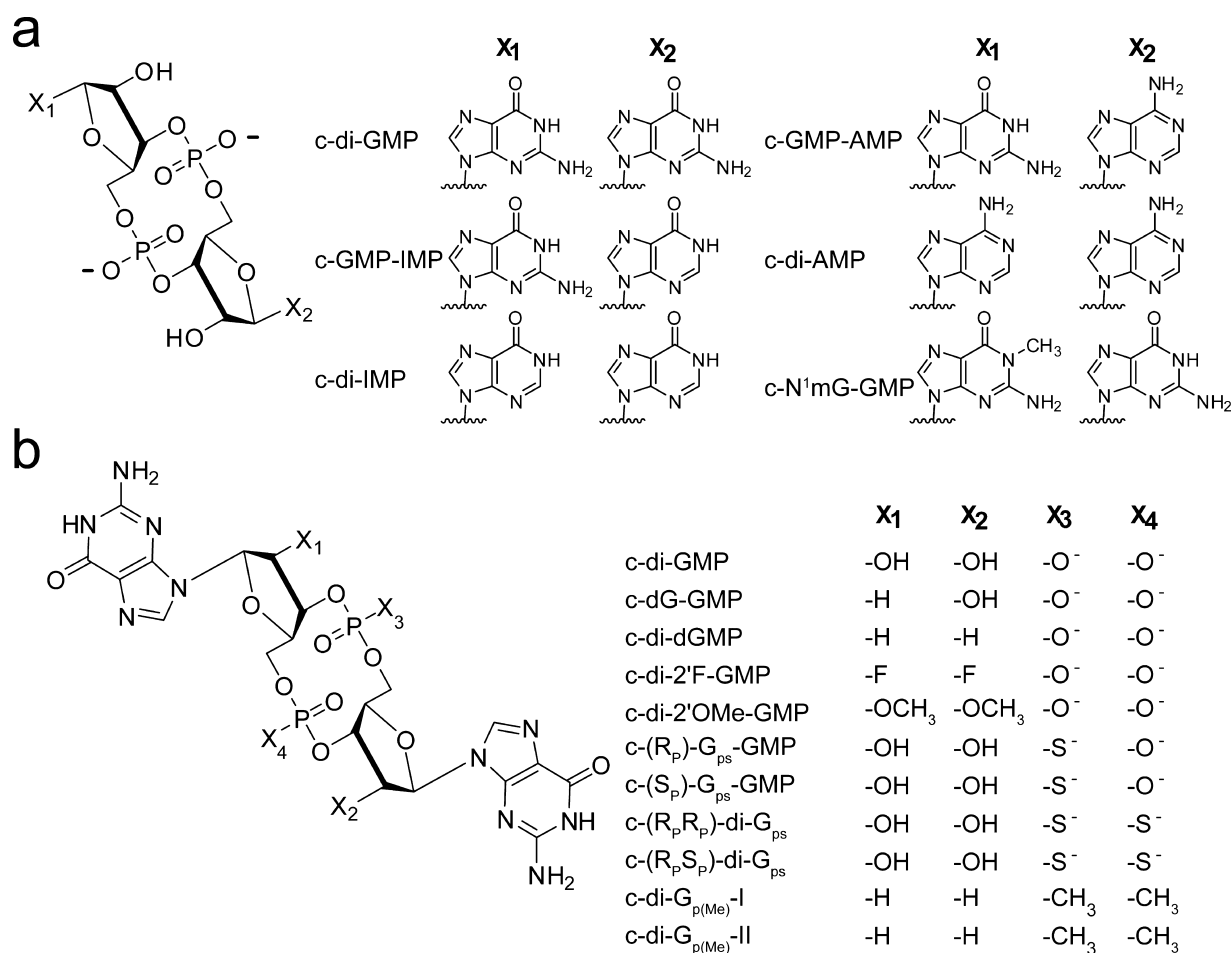


Figure 3. Structures of c-di-GMP analogues. (a) Base-modified c-di-GMP analogues. Guanine was replaced with inosine, adenosine, or *N*¹-methylguanosine as indicated by X₁ and X₂. (b) Ribose- and phosphate-modified analogues. Modifications made to the ribose rings are indicated by X₁ and X₂, and those made to the phosphate linkages are indicated by X₃ and X₄.

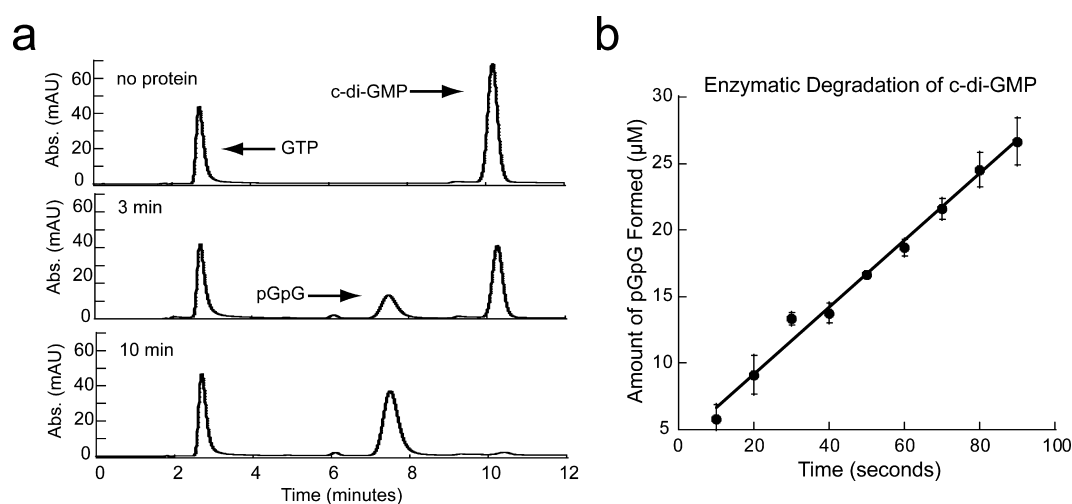


Figure 4. Monitoring the enzymatic degradation of c-di-GMP by HPLC. (a) HPLC traces showing elution of GTP, pGpG, and c-di-GMP. Peaks corresponding to each of these compounds are labeled. After 10 min under the experimental conditions used, c-di-GMP is completely degraded. (b) The amount of pGpG formed was measured in the linear range of the reaction to determine the initial velocity of c-di-GMP degradation. Data were collected and analyzed in the same manner for those analogues that were susceptible to degradation.

CC3396 to determine which guanine analogues are most tolerated by the EAL domain (Figure 3a).

With the exception of c-di-AMP, all base-modified analogues tested were susceptible to EAL domain-catalyzed degradation

(Table 1). Second-messenger analogues containing either one inosine (c-GMP-IMP) or *N*¹-methylguanine (c-*N*¹mG-GMP) base in place of a single guanine were both degraded to their corresponding linear products at a rate within 2–3-fold of that

Table 1. Rates of Enzymatic Hydrolysis for c-di-GMP and Its Nucleotide Analogues by EAL Domain Phosphodiesterase Protein CC3396^a

modification	analogue	rate (k_{cat}) [mol min ⁻¹ (mol of enzyme) ⁻¹]	fold change
base ^c	c-di-GMP ^b	15 ± 1.6	—
	c-GMP-IMP	7.0 ± 0.4	2.1
	c-N ¹ mG-GMP	4.5 ± 0.7	3.3
	c-di-IMP	0.51 ± 0.06	29
	c-GMP-AMP	0.48 ± 0.04	31
ribose ^b	c-di-AMP	nd ^d	—
	c-dG-GMP	6.4 ± 0.2	2.3
	c-di-dGMP	1.1 ± 0.6	14
	c-di-2'-F-GMP	2.9 ± 0.4	5.2
	c-di-2'-OMe-GMP	0.9 ± 0.2	17
phosphate ^b	c-(S _p)-G _{ps} -GMP	29 ± 1.8	0.5
	c-(R _p)-G _{ps} -GMP	8.3 ± 0.6	1.8
	c-(R _p R _p)-di-G _{ps}	nd ^d	—
	c-(R _p S _p)-di-G _{ps}	nd ^d	—
	c-di-dG _{p(Me)} -I	nd ^d	—
	c-di-dG _{p(Me)} -II	nd ^d	—

^aAll data are the average of at least three independent trials. ^bRates were measured under saturating conditions of substrate and therefore correspond to k_{cat} . ^cRates measured under multiple-turnover conditions with 100 μ M substrate and 1 μ M enzyme. ^dNo degradation under the conditions tested after incubation with CC3396 for 24 h.

for the native second messenger. In contrast, both c-di-IMP and c-GMP-AMP were hydrolyzed approximately 30-fold slower than c-di-GMP (Table 1). However, for the base-modified analogues tested, we did not determine if the substrate concentration (100 μ M) was saturating, and it is likely that the observed decrease in rate for c-di-IMP and c-GMP-AMP is a binding effect rather than a catalytic effect. These data indicate that the EAL domain can accommodate structurally related analogues of guanine in the active site and suggest that specific recognition of only one guanine base is sufficient for binding and degradation of the second messenger.

In the HPLC profile of c-N¹mG-GMP after treatment with CC3396, two product peaks with retention times differing from that of the cyclic dinucleotide were observed (Figure S1a of the Supporting Information). These peaks likely correspond to the two possible linear products, p-N¹mG-pG and pGpN¹mG. They were formed in unequal amounts, suggesting that CC3396 preferentially binds the modified dinucleotide in one orientation. For c-GMP-IMP, a major product peak and a much smaller minor peak were observed, suggesting a preference for binding orientation for that analogue as well (Figure S1b of the Supporting Information). In contrast, only one product peak was observed for c-GMP-AMP (Figure S1c of the Supporting Information). This suggests there is a preference for binding orientation between the G and the A. These observations are consistent with structural analysis, which predicts that one guanine base is more extensively recognized than the other.^{23,25}

Enzymatic Hydrolysis of Analogues Containing Ribose Modifications. Based on structural analysis, no specific contacts are made to the 2'-OH groups of the c-di-GMP ribose sugars by the EAL domain (Figure 1b,c),^{23–25} suggesting that hydroxyl modification would not significantly affect the rate of catalysis. To test this prediction, we first measured the rate of degradation for the 2'-deoxy ligands, c-di-dGMP and c-dG-GMP (Figure 3b). We found no significant effect on degradation for c-dG-GMP (2-fold); however, c-di-

dGMP was hydrolyzed approximately 14-fold slower than c-di-GMP (Table 1), suggesting that modification of both ribose rings increases the resistance of the second messenger toward EAL domain-catalyzed degradation. To further test this observation, we measured the rate of hydrolysis for 2'-fluoro-modified (c-di-2'-F-GMP) and 2'-O-methyl-modified (c-di-2'-OMe-GMP) analogues (Figure 3). These ribose modifications have been shown to increase the stability of nucleic acids toward nucleases that cleave phosphodiester bonds.⁴⁰ We found that both analogues were hydrolyzed at a slower rate than c-di-GMP, with c-di-2'-F-GMP degraded 5-fold slower and c-di-2'-OMe-GMP degraded 17-fold slower (Table 1). The 2'-O-methyl substitutions have a slightly larger effect on hydrolysis than 2'-fluoro substitutions, but both substitutions were degraded with reasonably good efficiency.

Enzymatic Hydrolysis of Analogues Containing Phosphate Modifications. In the proposed mechanism for c-di-GMP hydrolysis by EAL domain proteins, both non-bridging oxygens of the scissile phosphate are contacted by an active site metal ion (Figure 1b).^{23,25} Furthermore, it was previously demonstrated that a c-di-GMP analogue containing one phosphorothioate linkage is hydrolyzed by membrane extracts possessing c-di-GMP specific PDE activity to only a single linear product, with the modified linkage at the internal phosphate of the dinucleotide product.³¹ This suggests that phosphorothioate linkages are resistant to enzymatic hydrolysis by EAL domain proteins and that a second-messenger analogue with phosphorothioate substitutions in place of both phosphodiester bonds may be resistant to enzymatic degradation. To investigate this hypothesis, we tested mono- and dithiophosphate c-di-GMP derivatives for degradation by the EAL protein.

On the basis of previous reports in the literature, we anticipated that the monothiophosphate c-di-GMP derivatives, c-(R_p)-G_{ps}-GMP and c-(S_p)-G_{ps}-GMP (Figure 3), would be efficiently degraded.³¹ As expected, both analogues were degraded at a rate within 2-fold of that for c-di-GMP (Table 1), confirming that modification to only one of the c-di-GMP phosphates does not significantly affect catalysis. Furthermore, only a single degradation product was detected by HPLC analysis for both the R_p and S_p derivatives, consistent with the previous observations that incubation of these analogues with membrane extracts containing c-di-GMP specific PDE activity also yielded only a single linear product.³¹

We next tested the ability of the protein to degrade the dithiophosphate analogues, c-(R_pR_p)-di-G_{ps} and c-(R_pS_p)-di-G_{ps} (Figure 3). In contrast to the monothiophosphate derivatives, we observed no detectable degradation of either dithiophosphate analogue even after a 24 h incubation with enzyme (Figure 5a,b). Even if 2% of the compound was hydrolyzed after 24 h, which is the limit of our detection, the rate of hydrolysis would be estimated to be 0.001 mol min⁻¹ (mol of enzyme)⁻¹, which is at least 12000-fold slower than that of c-di-GMP. This indicates that substitution of either phosphate oxygen of the scissile phosphate with sulfur renders the second messenger highly resistant to EAL domain-catalyzed degradation. To ensure that the linear hydrolysis product was not coeluting with the cyclic starting material, we collected the peak and analyzed it by ESI-MS to determine the mass of the corresponding compound. For both diastereomers, the mass of the cyclic, but not the linear, product was detected (Table S2 of the Supporting Information).

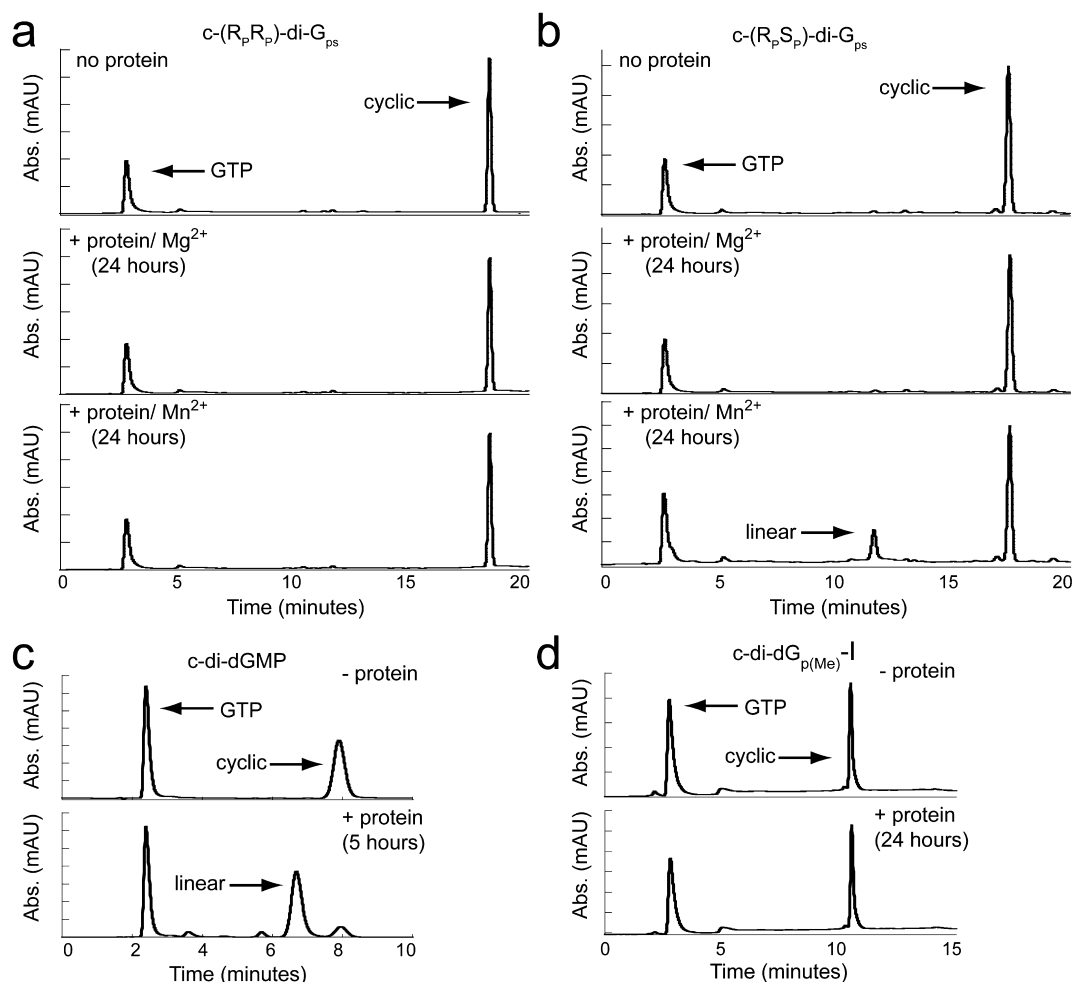


Figure 5. HPLC traces monitored at 254 nm showing degradation of backbone-modified analogues by EAL domain protein CC3396. Peaks corresponding to GTP, cyclic, and linear dinucleotides are labeled. (a) In the absence of protein (top), a single peak corresponding to $c-(R_pR_p)\text{-di-G}_{ps}$ is observed. Only a peak corresponding to the cyclic dinucleotide is detected after a 24 h incubation with the protein in the presence of either 10 mM $MgCl_2$ (middle) or 10 mM $MnCl_2$ (bottom). (b) HPLC analysis of $c-(R_pS_p)\text{-di-G}_{ps}$ in the absence (top) and presence (middle and bottom) of protein. No degradation product is observed with $MgCl_2$, whereas a peak corresponding to the linear product is observed in the presence of $MnCl_2$. Approximately 20% of the cyclic product is hydrolyzed after 24 h. (c) HPLC traces of the $c\text{-di-dGMP}$ degradation reaction. The top panel shows the reaction mixture in the absence of protein and the bottom panel in the presence of protein. After a 5 h incubation with CC3396, $c\text{-di-dGMP}$ is almost completely hydrolyzed (>85%). (d) HPLC traces of $c\text{-di-dG}_{p(Me)}\text{-I}$ in the absence (top) and presence (bottom) of CC3396. No hydrolysis products can be detected even after a 24 h incubation with the protein. The same lack of reactivity was observed for $c\text{-di-dG}_{p(Me)}\text{-II}$ after treatment with the protein for 24 h (data not shown).

Because nuclease activity against phosphorothioate-containing nucleic acids is often recovered in the presence of manganese,^{41,42} we next tested both dithiophosphate analogues for degradation by the EAL domain PDE with 10 mM manganese in place of 10 mM magnesium in the reaction buffer. Under these conditions, we found that $c-(R_pS_p)\text{-di-G}_{ps}$ was approximately 20% degraded after 24 h and estimated that the rate of degradation was approximately 1000-fold slower than that of $c\text{-di-dGMP}$ (Figure 5b). In contrast, no degradation of $c-(R_pR_p)\text{-di-G}_{ps}$ was observed under these conditions (Figure 5a). This indicates that manganese is able to rescue the activity of the protein, but only for one of the phosphorothioate diastereomers. Taken together, these data demonstrate that replacing both phosphodiester linkages with phosphorothioate linkages renders the second messenger highly resistant to EAL domain-catalyzed degradation.

Synthesis and EAL Domain-Catalyzed Degradation of a Backbone Neutral $c\text{-di-GMP}$ Analogue. The enzymatic degradation studies of the dithiophosphate $c\text{-di-GMP}$ deriva-

tives suggest that modification of both phosphate linkages renders the second messenger resistant to degradation by EAL domain proteins. To further explore this observation, we synthesized a second backbone-modified $c\text{-di-GMP}$ derivative containing methylphosphonate linkages in place of both canonical phosphodiester bonds. The methylphosphonate derivative of $c\text{-di-GMP}$, termed $c\text{-di-dG}_{p(Me)}\text{-I}$, substitutes a nonbridging phosphate oxygen of each phosphodiester linkage with an isosteric methyl group, resulting in a molecule with a neutral backbone (Figure 3). Because the methylphosphonamidite monomers are only commercially available in the 2'-deoxy version because of the instability of ribo-methylphosphonate linkages,^{43–45} we synthesized the di-2'-deoxy derivative and tested this molecule in comparison to the di-2'-deoxy parental dinucleotide with standard phosphate linkages, $c\text{-di-dGMP}$ (Figure 3). We modified the solid phase method for $c\text{-di-GMP}$ reported by Kiburu et al. to access this $c\text{-di-GMP}$ derivative.³⁸ We took advantage of the fact that methyl phosphonamidite starting monomers do not possess any

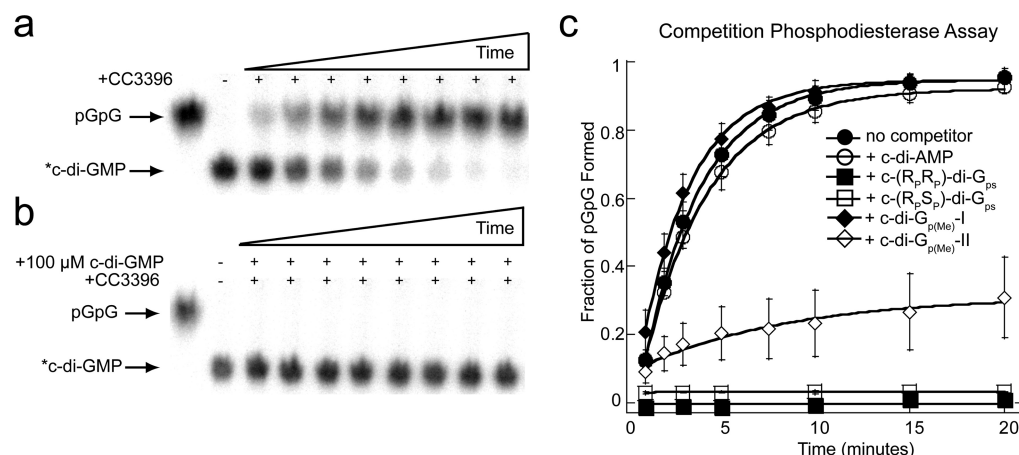


Figure 6. Degradation of radiolabeled c-di-GMP ($^*\text{c-di-GMP}$) in the presence of EAL domain resistant c-di-GMP analogues. (a) PEI-cellulose TLC showing degradation of $^*\text{c-di-GMP}$ to pGpG by CC3396 over time. (b) Degradation of $^*\text{c-di-GMP}$ in the presence of unlabeled c-di-GMP. (c) Fraction of pGpG formed from degradation of $^*\text{c-di-GMP}$ by CC3396 over time in the presence of unlabeled c-di-AMP, dithiophosphate, and methylphosphonate analogues.

phosphate protecting groups. This allowed for solid phase dinucleotide synthesis, followed by selective cleavage of the linear dimer from the solid support under mild basic conditions without the removal of any phosphate protecting groups, or the protecting groups on the bases (Figure 2a). After both guanosine methyl phosphonamides had been coupled to the solid support, the linear dinucleotide was cleaved from the bead via β -elimination under basic conditions and cyclized in solution.

Replacing one of the nonbridging oxygens on each phosphate renders both phosphates chiral and results in three different possible diastereomers (R_pR_p , R_pS_p , or S_pS_p). The fourth diastereomer (S_pR_p) is equivalent to the R_pS_p isomer because of the 2-fold symmetry within the cyclic dinucleotide. Using solid phase chemistry as described above, we were able to synthesize and isolate two of the three possible diastereomers after HPLC purification (Figure 2b). This is similar to what was achieved for the synthesis of the phosphorothioate analogues. In that case, the cyclization reaction proceeded with stereochemical specificity.^{35,46} Thus, the absence of the third isomer is likely due to the cyclization step producing a linkage of only one chirality [most likely R_p (see Discussion)]. Because of the small amount of product generated from solid phase synthesis, we were not able to make enough material to assign absolute stereochemistry. We have designated these analogues c-di-dG_{p(Me)}-I and c-di-dG_{p(Me)}-II based on their order of elution during HPLC purification (Figure 2b). However, tentative stereochemical assignments are possible on the basis of their relative affinities to riboswitch targets [binding data suggest isomer I is R_pR_p and isomer II is R_pS_p (see Discussion)].

To determine if methylphosphonate c-di-GMP derivatives are nuclease resistant, we tested both diastereomers for degradation by the EAL domain protein CC3396. Similar to what we observed for the phosphorothioate derivatives, neither analogue was degraded even after a 24 h incubation period with the protein (Figure 5d). In contrast, the parental dinucleotide, c-di-dGMP, is nearly completely hydrolyzed (>85%) after 5 h (Figure 5c). Assuming the maximal possible level of degradation of 2% after 24 h, the rate of degradation is more than 1000-fold slower than that of the parental dinucleotide [$1.1 \text{ mol min}^{-1} (\text{mol of enzyme})^{-1}$ (Table 1)]. Methyl substitution of both phosphodiester bonds results in a c-di-

GMP analogue that has high resistance to EAL domain-catalyzed degradation.

We next sought to determine if the EAL domain resistance of the dithiophosphate and methylphosphonate c-di-GMP analogues results from a weaker affinity for the protein for these analogues or if the protein is unable to perform catalysis on these modified phosphate centers. To differentiate between these two possibilities, we looked at the ability of these analogues to competitively inhibit the degradation of radiolabeled c-di-GMP ($^*\text{c-di-GMP}$).⁴⁷ In the presence of trace amounts of $^*\text{c-di-GMP}$ and a large excess of an unlabeled competitor analogue that is able to bind the EAL domain, $^*\text{c-di-GMP}$ binding and degradation should effectively be blocked. We found that in the absence of any competitor analogue, $^*\text{c-di-GMP}$ was completely degraded to pGpG after approximately 15 min, whereas in the presence of a large excess of unlabeled c-di-GMP, no degradation was observed as expected (Figure 6a,b). However, in the presence of 100 μM c-di-AMP, which does not bind the EAL domain, we found that $^*\text{c-di-GMP}$ is degraded at nearly the same rate as in the absence of a competitor (Figure 6c). This establishes that only molecules able to bind the EAL domain can prevent $^*\text{c-di-GMP}$ degradation under these conditions.

To determine if the dithiophosphate analogues bind the EAL domain, we tested their ability to block degradation of $^*\text{c-di-GMP}$ as described above. In the presence of either 100 μM c- (R_pR_p) -di-G_{ps} or c- (R_pS_p) -di-G_{ps}, we found that $^*\text{c-di-GMP}$ degradation was completely inhibited (Figure 6c). This indicates that the EAL domain can recognize and bind both dithiophosphate diastereomers and suggests that their nuclease resistance results from the inability of the protein to catalyze hydrolysis on these phosphate-modified substrates. We estimated the affinity (K_i) of both the R_pR_p and R_pS_p diastereomers for the EAL domain to be within 10-fold of the K_m of c-di-GMP (measured as $\sim 110 \text{ nM}$), indicating that these analogues may function as effective competitive inhibitors of this enzyme class [$K_i \sim 480 \text{ nM}$ for c- (R_pR_p) -di-G_{ps}; $K_i \sim 820 \text{ nM}$ for c- (R_pS_p) -di-G_{ps}].

For the methylphosphonate derivatives, we found that isomer I was unable to block second-messenger degradation even though the analogue concentration was approximately 6 orders of magnitude greater than that of $^*\text{c-di-GMP}$. This

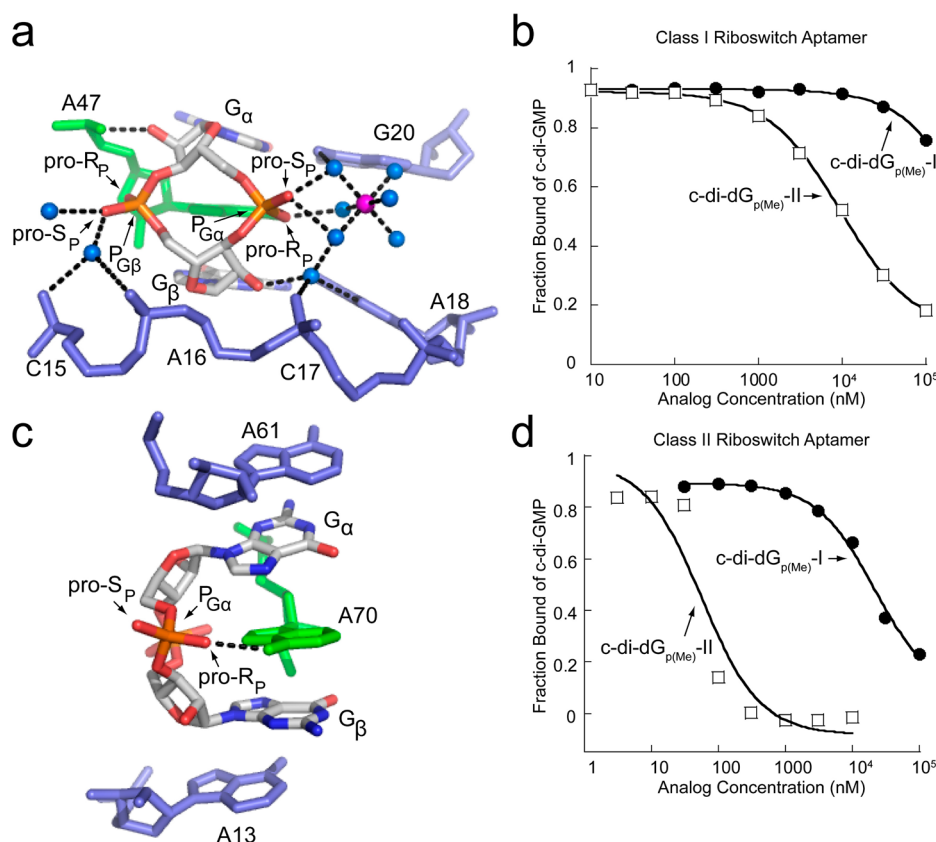


Figure 7. Backbone recognition of c-di-GMP and binding of methylphosphonate analogues by the class I and class II riboswitch aptamers. (a) c-di-GMP bound to the class I aptamer (PDB entry 3MXH). c-di-GMP is colored by atom: white for carbon, blue for nitrogen, red for oxygen, and orange for phosphorus. The highly conserved adenosine that intercalates between the guanine bases of c-di-GMP is colored green, and additional RNA residues are colored blue. Water molecules are shown as blue spheres and magnesium ions as purple spheres. The *pro-R_P* and *pro-S_P* oxygens are indicated to show the sites of methyl substitutions. Hydrogen bonds are indicated by dashed lines. (b) Binding of methylphosphonate derivatives to the class I riboswitch. The c-di-dG_{p(Me)}-I binding curve is shown (●), and the c-di-dG_{p(Me)}-II curve is shown (□). (c) c-di-GMP bound to the class II aptamer (PDB entry 3Q3Z). The coloring and labeling are the same as in panel a. (d) Binding of the methylphosphonate derivatives to the class II riboswitch. The labeling is the same as in panel b.

Table 2. K_d Measurements of Methylphosphonate Analogues for the Class I and Class II c-di-GMP Binding Riboswitches Determined by the Competition Gel Shift Assay

analogue	class I			class II		
	K_d (nM)	fold change ^a	$\Delta\Delta G_{bind}$ (kcal/mol) ^a	K_d (nM)	fold change ^a	$\Delta\Delta G_{bind}$ (kcal/mol) ^a
c-di-dGMP ^b	2600 ± 88	—	—	11 ± 1.2	—	—
c-di-GMP ^b	1.4 ± 0.1	[1900] ^c	−4.4	2.2 ± 0.2	[5]	−0.9
c-di-dG _{p(Me)} -I	>100000	>38	>2.2	1000 ± 93	91	2.7
c-di-dG _{p(Me)} -II	540 ± 57	[5]	−0.9	2.4 ± 0.3	[5]	−0.9
c-(<i>R_PR_P</i>)-di-G _{ps} ^b	150 ± 33	[17]	−1.7	3.6 ± 0.6	[3]	−0.7
c-(<i>R_PS_P</i>)-di-G _{ps} ^b	750 ± 63	[3]	−0.7	4.0 ± 0.8	[3]	−0.6

^aCalculated relative to the parental dinucleotide c-di-dGMP. ^bBinding data for these compounds were previously reported.³⁴ ^cBrackets indicate an increase in the level of binding.

indicates that c-di-dG_{p(Me)}-I has very little affinity for the EAL protein. For isomer II, we observed that only 30% of c-di-GMP was degraded to pGpG under the same conditions in which 95% of the substrate was degraded in the absence of any competitor analogue (Figure 6c). This indicates that this analogue retains some affinity for the EAL domain. On the basis of the affinity measurements for the dithiophosphate analogues as indicated above along with the competition studies indicating that isomer II retains some affinity for the EAL domain, we estimate the K_i for isomer II to be >10-fold greater than the K_m of c-di-GMP. Taken together, these data suggest

that for isomer I, EAL domain resistance is primarily due to the inability of this compound to bind the protein whereas for isomer II, resistance stems from a combination of both weaker binding and chemical effects.

Effects of Methylphosphonate Substitutions on c-di-GMP Binding by the Class I and Class II Riboswitches. We previously published a structure–function study of c-di-GMP analogue binding to both the class I and class II riboswitches.³⁴ With the exception of the novel methylphosphonate derivative developed here, all other analogues tested for EAL domain susceptibility were tested for riboswitch binding,

indicating which analogues with increased enzymatic resistance still retain affinity for a known c-di-GMP target. Therefore, we next sought to determine the effects of methylphosphonate substitution of c-di-GMP on the affinity for its second-messenger riboswitches.

Structural characterization of binding of c-di-GMP to the class I and class II riboswitches has revealed that class I riboswitches more extensively recognize the ribosyl phosphate backbone than class II riboswitches^{26,27} (Figure 7a,c). We measured the affinity (K_d) of both methylphosphonate diastereomers for the class I riboswitch (Figure 7b). For c-di-dG_{p(Me)}-I, only slight binding was detectable at high analogue concentrations, suggesting that the K_d is $>100 \mu\text{M}$ (Table 2). In contrast, this RNA bound the second diastereomer, c-di-dG_{p(Me)}-II, with surprisingly good affinity (540 nM). This is 5-fold tighter than the binding of the parental dinucleotide c-di-dGMP (Table 2) and indicates that binding of c-di-dG_{p(Me)}-II resulted in improved affinity [-0.9 kcal/mol (Table 1)] relative to that of the di-deoxy cyclic dinucleotide analogue.

Similar studies were performed on the class II riboswitch aptamer (Figure 7d). This RNA bound c-di-dG_{p(Me)}-I significantly more weakly than c-di-dG_{p(Me)}-II. The K_d of isomer I was approximately $1 \mu\text{M}$, 90-fold weaker than that of c-di-dGMP (Table 2). As with class I, we found that the c-di-dG_{p(Me)}-II analogue bound the class II RNA 5-fold tighter than c-di-dGMP with a K_d nearly identical to that of c-di-GMP [2.4 nM (Table 2)]. For both riboswitch classes, an increase of 0.9 kcal/mol upon binding of isomer II was observed relative to that of the parental double deoxy analogue (Table 2).

DISCUSSION

The enzymes responsible for the metabolism of c-di-GMP regulate the intracellular concentration of the second messenger, which ultimately controls diverse behaviors in bacteria, including biofilm formation, motility, and virulence response.^{2,6,9,10} Here, we probed the specificity of c-di-GMP specific EAL domain PDE CC3396 for its cognate ligand to identify analogues that are resistant to enzymatic hydrolysis. We found that modification of both GMP units of c-di-GMP is necessary to significantly affect the stability of the compound to hydrolysis. Substitution of both phosphates with either phosphorothioate or methylphosphonate linkages yields an EAL domain resistant second-messenger analogue. In addition, one of the two methylphosphonate diastereomers synthesized and tested (isomer II) had increased affinity for the c-di-GMP riboswitches that function as downstream effectors in this signaling pathway relative to its parental dinucleotide, c-di-dGMP. These results suggest that modifying both c-di-GMP phosphate linkages is a general strategy for designing second-messenger analogues that are stable to EAL domain-catalyzed degradation yet can still bind known downstream RNA effectors in this signaling pathway.

c-di-GMP has a 2-fold axis of symmetry, and the second messenger can bind the active site of the EAL domain in two possible orientations, with the modified GMP moiety placed either distal (P2) or proximal (P1) to the site where catalysis is taking place (Figure 1b). As a result, for asymmetric ribose- or phosphate-modified second-messenger analogues in which one GMP moiety remains unmodified, the rate of enzyme-catalyzed degradation by the EAL domain PDE was not significantly affected. For example, monothiophosphate and single 2'-deoxy-modified ligands are degraded by the PDE at nearly the same rate as the native second messenger, whereas the doubly

modified ligands with these same substitutions are degraded significantly more slowly than c-di-GMP.

With the exception of c-di-AMP, modification to the guanine bases does not render the second messenger resistant to EAL domain-catalyzed degradation. Substitution of one c-di-GMP base with either inosine or N^1 -methylguanine, which both are close structural analogues of guanine, had little to no effect on the rate of degradation. An analogue with one adenosine substitution, c-GMP-AMP, was still recognized and degraded, whereas the doubly substituted analogue c-di-AMP was not. This suggests that specific recognition of only one c-di-GMP base is sufficient for binding to the PDE, albeit with weaker affinity if the second site is substituted.

Interestingly, c-GMP-AMP has been discovered as a new second messenger in *V. cholerae*, synthesized by a novel class of dinucleotide cyclases that preferentially produces this dinucleotide over c-di-GMP and c-di-AMP.⁵⁶ The ability of the EAL domain protein studied here to degrade c-GMP-AMP, as discussed above, suggests that specific recognition of only one guanine base is necessary for c-di-GMP recognition and subsequent degradation. However, sequence alignments of EAL domain proteins indicate that guanine base recognition is achieved primarily through highly conserved residues involved in G_2 binding, whereas those residues involved in G_1 binding are highly variable (Table S1 of the Supporting Information). Therefore, in organisms that utilize both c-di-GMP and potentially c-GMP-AMP, such as *V. cholerae*, recognition of G_1 by endogenous EAL domain phosphodiesterases may be more stringent to specifically discriminate against c-GMP-AMP.

Dithiophosphate c-di-GMP analogues can bind the EAL domain, but what is the molecular basis of their nuclease resistance? Crystal structures of active EAL domain proteins reveal that two metal ions are bound in the active site and that both coordinate the proposed nucleophilic water molecule.^{23,25} In addition, each metal ion coordinates a different nonbridging phosphate oxygen of the c-di-GMP scissile phosphate [P1 (Figure 1b)]. This is different than the classic two-metal ion nucleases in which both metal ions coordinate the same phosphate oxygen⁴⁸ and often show stereochemical specificity for degrading phosphorothioate linkages (R_p or S_p).^{49–51} For many systems, substituting sulfur for oxygen at the site of metal coordination destroys enzymatic activity.⁵² In the case of c-di-GMP, sulfur substitution may simply displace one of the active site metals, significantly affecting the ability of the protein to hydrolyze the scissile phosphate linkage. However, unlike some systems in which one oxygen is strongly affected by sulfur substitution and the other is not,^{49–51} both oxygens show strong effects for c-di-GMP degradation. Although dithiophosphate analogues are resistant to enzymatic hydrolysis, these derivatives retain the ability to bind the EAL domain with reasonable affinity and therefore have the potential to function as competitive inhibitors of these c-di-GMP-metabolizing enzymes.

Similar to phosphorothioate substitution of c-di-GMP, methylphosphonate substitution also rendered the second messenger resistant to EAL domain-catalyzed degradation. However, the methylphosphonate analogues were weaker inhibitors of the EAL domain than the phosphorothioate derivatives. In contrast, both the phosphorothioate- and methylphosphonate-substituted derivatives of c-di-GMP bound both the class I and class II c-di-GMP riboswitches with good affinity, and in the case of one methylphosphonate diastereomer, with an affinity better than the parental analogue.

This suggests that these enzymatically stable versions of second-messenger analogues may be able to persist longer in the cell and may therefore be useful chemical tools for probing RNA-mediated c-di-GMP signaling *in vivo*.

While the absolute stereochemistry of the methylphosphonate derivatives is unknown, the data are consistent with the possibility that c-di-dG_{p(Me)}-I is the R_pR_p diastereomer based on the affinity for riboswitch targets. The interaction between the nonbridging phosphate oxygen of P_{Gα} and the adenosine that intercalates between the c-di-GMP bases is the only conserved backbone contact between the otherwise distinct modes of ligand recognition by the two classes of riboswitches²⁷ (Figure 7a,c). This is the only phosphate contact made to c-di-GMP by the class II riboswitch.²⁷ Therefore, substitution of the *pro*-R_p oxygen of P_{Gα} is expected to have negative effects on binding for both RNAs. Because methyl groups are isosteric with oxygen and the *pro*-R_p oxygen of P_{Gβ} is not recognized by either riboswitch, replacing this phosphate oxygen with a methyl group is not expected to have a significant impact on ligand affinity. The R_pS_p derivative can bind in two possible orientations, one in which the highly conserved contact to the *pro*-R_p oxygen of P_{Gα} is not affected. In contrast, the R_pR_p diastereomer retains the 2-fold symmetry axis and presents a *pro*-R_p oxygen substitution at P_{Gα} in both binding orientations. The large loss of affinity for c-di-dG_{p(Me)}-I by both the class I and class II riboswitches suggests that this contact is being disrupted and that isomer I is the R_pR_p isomer. Because of the stereochemical specificity of the cyclization step, only two of the three possible diastereomers are produced, and each must contain at least one R_p linkage. This suggests that c-di-dG_{p(Me)}-II is the R_pS_p isomer.

While methylphosphonate modification of both c-di-dGMP phosphate linkages had negative effects on binding and catalysis by the EAL domain, positive effects on binding by riboswitch targets of the second messenger were observed. The affinity of one of the methylphosphonate diastereomers [c-di-dG_{p(Me)}-II] for the class II riboswitch was almost identical to that of c-di-GMP and tighter than that of its parental dinucleotide, c-di-dGMP. This suggests that this analogue could affect the function of its RNA target molecule to the same extent as the native second messenger, while remaining more stable to degradation by EAL domain PDEs. Although the other methylphosphonate diastereomer [c-di-dG_{p(Me)}-I] bound to the class II riboswitch with weaker affinity, the EAL domain was unable to recognize this compound even at very high analogue concentrations. Thus, this EAL domain resistant c-di-GMP derivative shows a strong preference for binding the class II c-di-GMP riboswitch over the EAL domain and could be useful for specifically perturbing RNA-mediated signaling processes without affecting the function of phosphodiesterases.

The class I riboswitch aptamer is also able to bind c-di-dG_{p(Me)}-II tighter than the parental dinucleotide c-di-dGMP, despite the significant number of contacts made to the phosphate backbone. Interactions with the 2'-hydroxyl groups make a large contribution to ligand binding by the class I riboswitch;³⁴ however, removal of this functional group in the context of the methylphosphonate substitution did not have as large an effect on binding as in the background of the native, charged molecule. This suggests that alleviating electrostatic repulsion between c-di-GMP and the riboswitch RNA increases the affinity of the second messenger for its RNA targets. In addition, previous studies have suggested that a large portion of the binding energy for the c-di-GMP riboswitches results from

base stacking interactions.^{34,53} Collectively, these observations indicate that it may be possible to completely replace the ribosyl phosphate backbone of the second messenger with neutral linkers and maintain affinity for riboswitch targets, provided that the guanine bases are held at the appropriate distance to maintain these high-affinity stacking interactions.

Although c-di-dG_{p(Me)}-II binds tighter to c-di-GMP riboswitches than the parental dinucleotide c-di-dGMP, the ribose version of this analogue is expected to have even higher affinity, particularly for the class I riboswitch. Although methylphosphonate linkages next to ribose hydroxyls are unstable,^{43–45} it may be feasible to synthesize the ribo-methylphosphonate derivative of c-di-GMP because of the limited conformational flexibility of this cyclic dinucleotide. Wang et al. recently reported the synthesis of a c-di-GMP derivative containing a sulfur substitution in place of a bridging phosphate oxygen that was stable at neutral pH.⁴⁷ When incorporated into linear RNA oligonucleotides, these linkages are highly labile because of in-line attack from the neighboring 2'-hydroxyl. However, when incorporated into c-di-GMP, this linkage proved to be stable because the cyclic backbone of the second messenger constrains the molecule and effectively prevents the 2'-hydroxyl from attacking the scissile phosphate.⁴⁷ Ribo-methylphosphonate linkages incorporated into c-di-GMP may be stable provided that the 2'-OH group is protected until after the cyclization reaction is complete. Improved synthetic methods will be necessary to fully explore the potential of this analogue as a chemical tool for studying c-di-GMP signaling.

Given that the EAL domain cannot efficiently hydrolyze second-messenger analogues with methylphosphonate or phosphorothioate substitutions, which closely resemble the canonical phosphodiester linkage, it is unlikely that a c-di-GMP derivative with a non-native backbone would be susceptible to enzyme-catalyzed degradation. Modifying the phosphate backbone provides the opportunity to increase not only the enzymatic stability of the second messenger but also its cell permeability. Neutral molecules often have increased cell permeability and therefore a greater chance of entering cells via passive diffusion as compared to charged compounds.^{54,55} Because c-di-GMP signaling is used by many pathogenic organisms to control lifestyle changes that often allow the bacteria to infect a host or form biofilms, which are highly resistant to current antibiotic treatments, the ability to target this pathway for therapeutic purposes is desirable. Thus, the degradation resistant phosphorothioate and neutral methylphosphonate derivatives studied here have desirable properties that may make them useful for studying c-di-GMP signaling *in vivo* and manipulating the biological processes under control of this second messenger.

■ ASSOCIATED CONTENT

● Supporting Information

Supplemental results, including two tables and two figures as described in the text. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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ABBREVIATIONS

c-di-GMP, bis-(3'-5')-cyclic dimeric guanosine monophosphate; PDB, Protein Data Bank.

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